

1 A review and statistical study of existing hysteresis models
2 for cementitious materials

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5 **Abstract**

6 Hysteretic behavior of water vapour sorption isotherms (WVSIs) has been widely rec-
7 ognized as one of main factors which can significantly affect moisture transport within
8 building materials exposed to natural weather conditions. Due to the lack of experi-
9 mental data for validation, there are few studies about hysteresis models in the field of
10 cementitious materials. In this paper, a detailed review of hysteresis models has been
11 presented. Models which have been initially developed for water transport in soils and
12 sands have been selected, including the conceptual and empirical models. The com-
13 prehensive comparisons with experimental data have been performed for cement pastes
14 and concretes. It enables to conclude a set of best models for prediction of WVSIs and
15 their hysteresis. Comparison results show that for the prediction of the first scanning
16 curves, available hysteresis models (either conceptual or empirical) perform similarly.
17 However, studied empirical models yield *pumping errors* for the secondary or higher
18 order scanning loops. A multi-level hysteresis modelling has been proposed as well.
19 Evaluations and recommendations for use of this multi-level modelling method have
20 been provided for an application purpose of compared models.

21 *Keywords:* cementitious materials, water vapour sorption isotherms, hysteresis,
22 scanning curves, statistics

23 **1. Introduction**

24 The durability of reinforced concrete structures and their service life are closely re-
25 lated to the simultaneous occurrence of many physical and chemical phenomena. These
26 phenomena are diverse in nature, but in common they are dependent on the moisture

27 properties of the material. Therefore, the prediction of the potential degradation of
28 cementitious materials requires the study of the movement of liquid-water and gas dif-
29 fusion in the material which is considered as a porous medium. For the modelling of
30 moisture transport in this kind of materials, such as the continuum model proposed
31 in [1], the WVSIs, describing the relationship between relative humidity RH (or capil-
32 lary pressure P_c) and water content θ (or degree of saturation S), are used to address the
33 equilibrium between liquid-water and vapour. A group of typical WVSIs are illustrated
34 in Fig. 1, which contains two main isotherms (adsorption AB and desorption BA) and
35 several scanning curves (the first and secondary wetting scanning curves CB and EB ,
36 and the first scanning curve in drying DE).

37 [Figure 1 about here.]

38 In experiments, only a limited number of WVSI loops can be measured. Never-
39 theless, in natural conditions, concrete structures undergo arbitrary drying and wetting
40 cycles. Thus, using relevant models to predict these curves for cementitious materials
41 is necessary for modelling of moisture transport.

42 A main characteristic of WVSIs is their hysteretic behavior, referring to the differ-
43 ent water content at the same RH value (see the differences between two main curves in
44 Fig. 1). The most studied reasons of moisture hysteresis in porous media, such as soils
45 and sands, are the “ink-bottle” effect (*e.g.*, [2]), the variation in liquid-solid contact
46 angle (*e.g.*, [3]) and the difference of spatial connectivity of pores during the drying
47 and wetting processes. For cementitious materials, the characteristics of the hysteretic
48 behavior might be different to soils or sands because cementitious materials contain a
49 wider range of magnitude of pore sizes (from gel pores to capillary pores) [4]. Experi-
50 ments showed that in the low RH range ($\leq 33\%$, according to [5]), hysteresis between
51 the desorption and adsorption isotherms is much smaller than in the high RH range.
52 This results from the nanostructure of C-S-H [6], which is not found in soils and sands.

53 Although pore systems of different types of materials are different, it is of interest
54 to test whether the hysteresis models from soil science are also applicable to cemen-
55 titious materials, as there are few studies on the hysteresis behavior of such materials.
56 Carlier *et al.* [7] evaluated different closed-forms of analytical WVSIs expressions used

57 in soil science. The authors concluded that all models were able to correctly represent
58 the desorption isotherms for mortars and concretes. Recently, Johannesson *et al.* [8]
59 and Derluyn *et al.* [9] have used hysteresis models to investigate hysteretic effects on
60 moisture behavior of cementitious materials.

61 The main objective of this paper is to properly describe WWSIs of cementitious ma-
62 terials, including their hysteresis, by using mathematical formulas. After a review of
63 the development of hysteresis models, we will focus on the comparison of commonly-
64 used WWSIs and hysteresis models that are mainly collected from soil science. Avail-
65 able experimental data in the literature for cement pastes and concretes will be used to
66 verify these models. The main features and recommendations will be given in the last
67 part according to a proposed multi-level modelling of hysteresis.

68 2. Review of hysteresis models

69 The first documented description of the relation $P_c-\theta$ in porous media was provided
70 by Haines [10]. After that, the hysteretic behavior of this relation was recognized and
71 several models had been developed to predict hysteresis. Those models can roughly be
72 classified into two groups: conceptual and empirical models [11, 12].

73 The conceptual models are mainly known as *independent* or *dependent* domain
74 models, which usually employs *distribution diagrams* to demonstrate the theory. They
75 assume that a domain is made up of groups of pores in a porous material. Poulavas-
76 silis was one of the first to adopt the independent domain models from magnetism
77 science to the research of water transport in porous media [13]. This theory, includ-
78 ing Preisach space and Néel's diagram, was developed and improved by Preisach [14],
79 Néel [15], Everett [16, 17], *etc.* However, it was reported that those models showed
80 discrepancies with experimental data [18–20]. Hence, researchers tried to modify and
81 develop new models for porous materials. Based on the *similarity hypothesis* proposed
82 by Philip [21], Mualem [22, 23] introduced the simplified independent domain models.
83 The basic idea of Mualem's independent models is that no interactions between those
84 pores are considered. In other words, each pore is independent of its neighbours. So
85 the drying or wetting of the pore system is only determined by the pore necks and pore

86 bodies. Two normalized variables, \bar{r} and $\bar{\rho}$ varying from 0 to 1, are used to represent
 87 the normalized radii of pore necks and pore bodies, respectively. When \bar{r} (or $\bar{\rho}$) = 1, it
 88 corresponds to the minimum capillary pressure P_c^{\min} and \bar{r} (or $\bar{\rho}$) is 0 when $P_c = P_c^{\max}$.
 89 At a given capillary pressure, θ can be obtained by integration of the pore water dis-
 90 tribution function $f(\bar{r}, \bar{\rho})$. In independent domain models, $f(\bar{r}, \bar{\rho})$ is formulated by the
 91 combined contribution of two independent pore water distribution functions $h(\bar{r})$ and
 92 $l(\bar{\rho})$, which describe the radii of pore necks and pore bodies, respectively.

$$\theta = \int \int f(\bar{r}, \bar{\rho}) d\bar{r} d\bar{\rho} = \int l(\bar{\rho}) d\bar{\rho} \int h(\bar{r}) d\bar{r} \quad (1)$$

93 A rectangle diagram was proposed by Mualem (see Fig. 2) [23] to depict how the
 94 independent domain model works. The horizontal axis represents the water distribution
 95 function $h(\bar{r})$ and the vertical axis represents the function $l(\bar{\rho})$. At equilibrium, the
 96 capillary pressure throughout the pore network is constant [24], either for pore necks
 97 or in pore bodies. So, $h(\bar{r})$ and $l(\bar{\rho})$ always change at the same capillary pressure, which
 98 is illustrated as the diagonal line in Mualem's diagram (see Fig. 2). Hence, the diagram
 99 is able to describe the total water distribution ($f(\bar{r}, \bar{\rho})$) through the product of $h(\bar{r})$ and
 100 $l(\bar{\rho})$. The water content at P_c during the wetting process is obtained by integrating the
 101 water distribution function from the smallest pore size (at P_c^{\max}) to current pore size
 102 (at P_c):

$$\theta_w(P_c) = \int_0^{\bar{\rho}(P_c)} l(\bar{\rho}) d\bar{\rho} \int_0^1 h(\bar{r}) d\bar{r} \quad (2)$$

103 To formulate water content clearly and to avoid using integral forms, two cumu-
 104 lative pore water distribution functions were introduced instead of the integrations for
 105 each axis [22].

$$L(P_c) = \int_0^{\bar{\rho}(P_c)} l(\bar{\rho}) d\bar{\rho} \quad H(P_c) = \int_0^{\bar{r}(P_c)} h(\bar{r}) d\bar{r} \quad (3)$$

106 According to Mualem's diagram, $L(P_c^{\max}) = 0$, $L(P_c^{\min}) = \theta^{\max}$ (maximum water
 107 content), and $H(P_c^{\max}) = 0$, $H(P_c^{\min}) = \theta^{\max}$. Hence, Eq. (2) can be rewritten as:

$$\theta_w(P_c) = \theta^{\max} L(P_c) \quad (4)$$

108 The expression of water content during the drying process is obtained in a similar
109 manner by an integration of Mualem's diagram.

$$\begin{aligned} \theta_d(P_c) &= \int_0^{\bar{\rho}(P_c)} l(\bar{\rho}) d\bar{\rho} \int_0^1 h(\bar{r}) d\bar{r} + \int_{\bar{\rho}(P_c)}^1 l(\bar{\rho}) d\bar{\rho} \int_0^{\bar{r}(P_c)} h(\bar{r}) d\bar{r} \\ &= \theta_w(P_c) + [\theta^{\max} - \theta_w(P_c)] H(P_c) \end{aligned} \quad (5)$$

110 From Eq. (5), we get the expression of H as follows:

$$H(P_c) = \frac{\theta_d(P_c) - \theta_w(P_c)}{\theta^{\max} - \theta_w(P_c)} \quad (6)$$

111 [Figure 2 about here.]

112 If both main isotherms are known ($\theta_d(P_c)$ and $\theta_w(P_c)$), functions H and L can be
113 determined easily by using Eqs. (4) and (6). Then, they are used to calculate scanning
114 loops.

115 The assumption of *independent* domain means that the pore systems are fully-
116 dried or fully-wet along with the changes of the pressure. Following the research of
117 Topp [19], Mualem and Dagan [25] adjusted the water distribution function (see Eq. 1)
118 by adding two weighting functions ($p_d \leq 1$ and $p_w \leq 1$).

$$\theta = \int \int p_d(\bar{r}, \bar{\rho}, \theta) p_w(\bar{r}, \bar{\rho}, \theta) f(\bar{r}, \bar{\rho}) d\bar{r} d\bar{\rho} \quad (7)$$

119 where p_d represents the volumetric ratio of the actually dried pores to pores supposed
120 to be dried in the independent domain theory. Similarly, p_w is the ratio for the wet-
121 ting process. Introducing these two weighting functions implies that the interactions
122 between pores and their neighbours have been considered. This is the basic idea of the
123 *dependent* domain models.

124 Through a comparison with experimental data, Mualem and Dagan [25] found that
125 p_d played the major role during both drying and wetting processes, while the effect

126 of p_w could be neglected ($p_w = 1$). Obviously, p_d is the key factor for Mualem's
127 dependent models. Various methods attempting to determine p_d have been reported in
128 the literature [26–28].

129 The empirical models, which have been applied to the modelling of moisture trans-
130 port in building materials (e.g., [8]), are developed based on fitting the shapes of ex-
131 perimental WVSIs. Normally, an empirical was proposed without physical meanings.
132 In the literature, there are a number of the empirical models, such as linear models
133 (e.g., [29]), slope models (e.g., [30]) and scaling-down models (e.g., [31]). Initially,
134 they were developed for one type of porous medium (such as soils and sands) and
135 not based on the physical representation of hysteresis. This paper has chosen recently
136 developed and easy-to-apply empirical models which can provide reasonable fitting re-
137 sults for experimental curves of cementitious materials. Such models usually have one
138 or two parameters that need to be determined from experimental data. For example,
139 Li [32] proposed an incremental relation to calculate hysteresis, with one parameter
140 controlling the shape of the curve. In the methods introduced by Wei *et al.* [33] and
141 Nyman *et al.* [34], parameters are used to initialise the slope of the scanning curves.

142 **3. Adopted comparison strategy**

143 A strategy of comparison of models for WVSIs and hysteresis to experimental data
144 is introduced. Statistical analysis tools will be presented.

145 *3.1. Procedures of comparisons*

146 According to the availability of experimental data, the normal procedure for the
147 calculation of hysteresis is like as follows:

- 148 • If the experimental data for both main adsorption and desorption branches
149 are measured, a model is needed to fit them, and then another model can
150 be used to predict scanning curves;
- 151 • If only one main branch is measured, the other one has to be determined
152 by means of a model, and then scanning curves can be predicted by another
153 model.

154 Based on above procedure, a multi-level modelling of hysteresis is proposed in Ta-
155 ble 1. It depends on the number of available experimental data. This number decreases
156 from Level 1 to Level 3. It may lead to different levels of prediction accuracy, which
157 will be revealed through comparisons in this study.

158 Table 1 indicates that a complete set of hysteresis models consists of three parts: ①
159 models to fit two main sorption isotherms; ② models to determine one main sorption
160 isotherm on the basis of the other measured main branch; ③ models to predict scan-
161 ning loops. In next sections, hysteresis models will be compared following these three
162 stages.

163 3.2. Data collection

164 In order to test the performance of the hysteresis models, measured WVSIs were
165 collected from the literature. The requirements for data collection are:

- 166 1) The collected data contain at least both main desorption and main adsorption isotherms.
167 It is better if some scanning curves have been measured.
- 168 2) The number of measurement data for each isotherm must be larger than the number
169 of undetermined parameters in models.

170 WVSIs can be measured by different methods. The saturated salt solution method
171 is a common gravimetric method in which the RH is controlled by saturated salt so-
172 lution in a small volume, such as in a desiccator. The mass of a specimen decreases
173 (during desorption) or increases (during adsorption) until equilibrium (mass stabiliza-
174 tion) is reached. Other methods, such as dynamic vapour sorption (DVS) [35, 36], use
175 a small specimen, usually less than 1 g, and a microbalance to monitor mass changes.
176 The advantage of this kind of instruments is that the entire desorption and adsorption
177 loop can be measured in one month or less, more rapidly than with the saturated salt
178 solution method, for which a method can take many months and even years. How-
179 ever, a potential problem with small specimens is that they cannot contain aggregates.
180 Measurements probably quantify only the sorption behavior for the paste in concrete
181 or mortar, rather than the global sorption behavior of the material. For instance, the in-
182 fluence of paste-aggregate interfacial transition zone on the sorption is not investigated
183 in this method. This may lead to different measurement results. Another risk is that

184 small specimens can be carbonated quickly during the preparation.

185 The studied experimental data are gathered in Table 2 from [5] measured by the
186 saturated salt solution method. These data can meet the above-mentioned comparison
187 requirements very well. All materials are made from the same OPC cement (CEM
188 I 52.5, according to EN 197-1 European standard). They include ordinary and high
189 performance hardened cement pastes and concretes. Water-to-cement ratios range from
190 0.2 to 0.6. High performance materials are Concrete2 and Paste2, which contain 10%
191 of silica fume to cement in weight. In order to reach chemical stability before the
192 measurements, materials were sealed curing for at least 6 months for concretes and 2
193 years for cement pastes. Then, cement pastes were crushed into small specimens with
194 masses of about 8 g. Because aggregates have no influence on the sorption process [5],
195 concrete specimens were cut into thin slices, 2 or 3 mm thickness with mass between 20
196 and 100 g. Measurements started from desorption at RH = 100%, decreasing step-by-
197 step to RH = 3%, considered as the dry reference state. Then, RH increased stepwise
198 to 100% to obtain the adsorption isotherm. To reduce the variations of RH during
199 the measurements as well as the carbonation risk, only a small hole on the top of the
200 desiccator is open for weighing. Besides, the equilibrium state for each RH step was
201 carefully defined as that the mass change is less than 0.001 g after one month. At least
202 three specimens were measured individually to determine one point on the sorption
203 curve.

204 3.3. Comparison criteria

205 For the fitting of the main sorption isotherms, all parameters in the models are
206 optimized to fit the experimental curves using a nonlinear least squared optimization
207 procedure, which aims to minimize the sum of the squares of the errors.

208 Two kinds of statistical criteria have been used to evaluate the accuracy of the mod-
209 elling results in the literature, the coefficient of determination (R^2) and a differential
210 residual-based error metrics (see [7, 11]). R^2 is defined as the ratio between the disper-
211 sion predicted by the model and the total dispersion of the measured data [11].

212 The accuracy of fitting (or prediction) is related to the number of parameters used in
213 a model. More parameters usually yields a higher R^2 value, but low robustness and low

214 sensitivity to parameters. To take the effect of the number of parameters into account,
 215 we use an adjusted R^2 instead of R^2 [7]:

$$R_{adj}^2 = \frac{(n-1)R^2 - (m-1)}{n-m} \quad (8)$$

216 where m is the number of parameters, n is the number of measured data. Equation (8)
 217 implies that $R_{adj}^2 < R^2$ when $m > 1$, $R_{adj}^2 = R^2$ when $m = 1$ and $R_{adj}^2 > R^2$ when $m = 0$.
 218 The latter case shows the advantage of using a non-undetermined parameter model.

219 The other statistical criterion is known as the normalized mean error (NME):

$$\text{NME} = \frac{1}{n} \sum_{i=1}^n \frac{|S_i^{pr} - S_i^{ms}|}{S_i^{ms}} \quad (9)$$

220 where S_i^{ms} and S_i^{pr} are measured and fitted (or predicted) values, respectively. Notice
 221 that the absolute value is used in Eq. (9). It can avoid the cancellation of positive and
 222 negative residuals. If the value of NME is close to 0, it indicates a better result.

223 3.4. Data pretreatment

224 WVSIs measured by experiments are generally RH– θ relation, but for calculation
 225 purpose, models usually need $P_c - \theta$ or $P_c - S$. To get a unified relation, the RH– S
 226 relation is used to display fitting or prediction results. The following equations are
 227 employed for transformations. From θ to S , the normalized equation is:

$$S = \frac{\theta - \theta_r}{\theta_{\max} - \theta_r} \quad (10)$$

228 where θ_r is the water content at dry reference state.

From RH to P_c , Kelvin's equation is used:

$$P_c = -\frac{\rho_l RT}{M_v} \ln \text{RH} \quad (11)$$

229 where ρ_l is liquid-water density ($\text{kg} \cdot \text{m}^{-3}$), $R = 8.314 \text{J} \cdot \text{K}^{-1} \cdot \text{mol}^{-1}$ is the gas constant,
 230 T is the absolute temperature (K). M_v is the molar mass of water molecule ($\text{kg} \cdot \text{mol}^{-1}$)
 231 and RH is a fraction in Eq. (11).

232 4. Main isotherms fitting models

233 Numerous models have been proposed to fit the main branches of WVSIs. Each
234 of these equations is applicable to one or several groups of materials (*e.g.*, sands and
235 soils). Models were selected in this research primarily based on whether they can fit
236 both desorption and adsorption isotherms for cementitious materials.

237 On the whole, two types of methods to fit WVSIs can achieve our purpose. They
238 are uni-modal and multi-modal models. As the name suggests, uni-modal models as-
239 sume that the material consists of one pore system. Thus, a simple single equation can
240 satisfactorily fit experimental data. In studies of Carlier *et al* [7], the Kosugi model
241 (K) [39] seems to be the better one for the fitting of the main desorption isotherm.
242 Besides, the van Genuchten's model [40], including two-parameter (VG2) and three-
243 parameter (VG3) versions, will be compared. The former version has been used in
244 studies of concretes [41, 42]. Two more models (Feng and Fredlund's model (FF) [43]
245 and Fredlund and Xing's model (FX) [44]) performing well in soil science will be val-
246 idated for cementitious materials. Multi-modal models consider that there are two or
247 even more pore systems in the material [45]. This consideration can improve the fitting
248 accuracy. A brief description of each selected model will be presented below.

249 4.1. The Feng and Fredlund's model (FF) [43]

250 The FF model is an empirical relationship which was used to fit desorption and
251 adsorption curves of a ceramic material [43]. A simple equation is used in this model.

$$S = \frac{a_F}{a_F + P_c^{b_F}} \quad (12)$$

252 Two parameters (a_F (Pa) and b_F) in Eq. (12) need to be determined. The research
253 performed in [46] showed that this curve-fitting equation is most applicable for low
254 swelling materials (*e.g.*, clay loam).

255 4.2. The van Genuchten's model [40]

256 Van Genuchten [40] proposed an *S*-shaped curve equation:

$$S = \left[1 + \left(\frac{P_c}{a_V} \right)^{n_V} \right]^{-m_V} \quad (13)$$

257 where a_V (Pa), m_V and n_V are three undetermined parameters. In addition, van Genuchten [40]
 258 suggested $m_V = 1 - 1/n_V$ to simplify Eq. (13) and a two-parameter equation was ob-
 259 tained.

260 4.3. The Fredlund and Xing's model [44] (FX)

261 When Fredlund and Xing [44] studied the pore size distribution on the basis of the
 262 VG2 model, they found that this model was not suitable in the high capillary pressure
 263 region. The authors introduced a modified pore size distribution function. Integration
 264 of this function yielded a new equation:

$$S = \left[\ln \left[e + \left(\frac{P_c}{a_X} \right)^{n_X} \right] \right]^{-m_X} \quad (14)$$

265 where e is the Euler number (a mathematical constant).

266 4.4. The Kosugi's model [39]

267 Using a lognormal pore size distribution, Kosugi [39] deduced the following ex-
 268 pression:

$$S = Q \left(\sigma^{-1} \ln \frac{P_c}{P_{cm}} \right) \quad (15)$$

269 where P_{cm} (Pa) is the capillary pressure related to the medium pore radius. σ is the
 270 standard deviation of log-transformed pore radii and is related to the width of the pore
 271 size distribution. If the pore size distribution is unknown, P_{cm} and σ have to be obtained
 272 by fitting experimental sorption isotherms. In Eq. (15), Q denotes the complementary
 273 normal distribution function, which is defined as:

$$Q(x) = (2\pi)^{-1/2} \int_x^{\infty} \exp\left(-\frac{x^2}{2}\right) dx \quad (16)$$

274 *4.5. The multi-modal model*

275 The generalized form of multi-modal models is written as:

$$S = \sum_{i=1}^N w_i S_i(P_c) \quad (17)$$

276 where N is the number of pore systems and $w_i \in [0, 1]$ ($\sum_i^N w_i = 1$) is a weighting
277 factor characterizing pore system i and w_i represents the water content contribution of
278 pore system i to the total water content in the material. If $N = 2$, Eq. (17) becomes
279 a bi-modal model. One well-known bi-modal model was proposed by Durner [45],
280 adopting VG2 equation for each pore system.

281 *4.6. Comparison results and discussion*

282 All materials in the collected database have been fitted by above models. Two sta-
283 tistical criteria R_{adj}^2 and NME for the fitting of the two main sorption isotherms are pro-
284 vided in Figs. 3 and 4. Results of two materials (Paste2 and Paste3, representing high
285 performance and ordinary materials) are illustrated in Figs. 6 and 5. Overall, all models
286 provide satisfactory fitting results (mean $R_{adj}^2 > 0.98$). The bi-modal model [45] shows
287 better fitting results than uni-modal models. For all models, the fitting of the adsorption
288 isotherm is better than the fitting of the desorption isotherm.

289 [Figure 3 about here.]

290 [Figure 4 about here.]

291 Concerning uni-modal models, they show a non-physical behavior since the adsorp-
292 tion curves intersect the desorption curves at the low RH (around 0.05 in two examples)
293 for some materials (see the enlarged figures in Figs. 5 and 6 for Paste2 and Paste3, re-
294 spectively). This phenomenon is not consistent with the experimental data. But for the
295 FX and VG3 models, this behavior is not as significant as for the other two-parameter
296 models (see the enlarged figures in Fig. 6). Both these models show better fitting results
297 than other uni-modal models (see values of R_{adj}^2 and NME in Figs. 3 and 4), either in
298 desorption or in adsorption isotherms. This suggests that these three-parameter models
299 can be used for a variety of cementitious materials. Moreover, one can remark that the

300 widely-used VG2 model [40] does not perform better than other models (see Figs. 3
301 and 4). Results also show that no selected model can simulate the sudden drop of the
302 desorption isotherms from 44% to 33% RH (see Fig. 6). This drop may be due to
303 measured water content in capillary pores changing to in C-S-H pores as emphasised
304 in reference [5].

305 [Figure 5 about here.]

306 Mean values of R_{adj}^2 for the bi-modal model are higher than 0.99 for both des-
307 orption and adsorption isotherms and values of NME are close to 0. For Paste3, the
308 bi-modal model roughly shows the same shape as the ones described with other models
309 (see Fig. 6a). For Paste2, this model performs much better in higher RH range, while
310 the predicted desorption isotherm does not reach $S=0$ at $RH=0$. This is clearly an
311 non-physical fitting result (see Fig. 5a). The result is likely caused by the very fine mi-
312 crostructure of the high performance cement pastes (*e.g.*, Paste2) and concretes (*e.g.*,
313 Concrete2). Because those materials can hold much water in the low RH range, the
314 measured points in the low RH range are higher compared to the ordinary materials,
315 such as the point at $RH=12\%$ in Fig. 5. This leads the non-physical fitting. Thus, it is
316 worth fitting WVSIs by using higher modal models.

317 A comparison of uni-, bi- and tri-modal models is made in Fig. 5. The performance
318 of the tri-modal model, nearly passing through all measured points, is obviously better
319 than the uni- and the bi-modal models for both main isotherms in the low RH range. In
320 terms of statistical analysis, using a tri-modal model improves the fitting results greatly
321 (see highlighted values of R_{adj}^2 and NME for Concrete2 and Paste2 in Figs. 3 and 4).
322 In the high RH range, the tri-modal model yields the similar shape of curve to the
323 bi-modal model.

324 [Figure 6 about here.]

325 However, the number of parameters in multi-modal models may be several times
326 higher than in uni-modal models. This results in a more difficult fitting procedure and
327 can even lead to instability. However, uni-modal models show acceptable fitting results

328 for the selected database, except for high performance materials. The choice of WVSI
329 model is thus highly dependent on the complexity of the measured isotherm.

330 Above, some commonly used WVSI models have been compared to measured
331 isotherms. In many cases, these models do not have a clear physical meaning. The
332 models proposed by Fredlund and Xing [44], van Genuchten [40] and Brutsaert [38]
333 originally came from Brooks and Corey's power type function [37], in which two pa-
334 rameters are involved, the air-entry pressure and a dimensionless parameter referring to
335 the pore-size distribution [37]. When van Genuchten developed the closed form equa-
336 tion, the introduced parameter a_V was disconnected to the air-entry pressure [47]; the
337 same is true for a_X in the FX model. So, a_V and a_X should be only seen as the air-entry
338 pressure related parameter. In the same way, n_V , m_V , n_X and m_X are parameters related
339 to the pore size distribution. The Kosugi's model [39] was developed for the soil and
340 two parameters (P_{cm} and σ) have to be determined based on the measured pore size
341 distribution. Due to the lack of the experimental data of the pore size distribution for
342 cementitious materials, the fitted P_{cm} and σ in this model cannot be verified whether
343 they still retain the same physical meaning as when they were first introduced. The
344 FF model is a pure empirical equation for the purpose of fitting the experimental data,
345 without any physical meaning being mentioned.

346 To conclude the above comparisons, this paper recommends that uni-modal models,
347 FX and VG3, can be used as a first tentative fitting. If one does not get satisfying
348 fitting results, multi-modal models can be used. In next sections of this paper, the FX
349 model will be used to fit the main sorption isotherms because it shows a slightly better
350 performance to fit desorption isotherms than the VG3 model (see Fig. 3).

351 **5. Prediction of one main isotherm from the other main branch**

352 Most hysteresis models require at least both main curves to predict scanning loops.
353 Nonetheless, measuring both sorption isotherms for cementitious materials is time
354 consuming and quite costly. In practice, it is still under debate whether the initial
355 fully dried state will change the microstructure during measurements of absorption
356 isotherms. Instead, the desorption isotherm is easier to determine from an initially

357 saturated state, which is also closer to the condition after the concrete structure form-
 358 works are removed. From this point of view, only the models to predict the adsorption
 359 isotherm from the desorption isotherm are evaluated herein.

360 In the literature, some comparisons have been done for sands and soils. Pham *et*
 361 *al.* [11] compared and ranked five selected models. Feng and Fredlund's model [43]
 362 (improved by Pham *et al.* [46]) appeared to be the most accurate one, followed by
 363 Mualem Model IV [48]. Meanwhile, in Maqsoud's comparison [12], Parlange's model [49]
 364 (modified by Braddock *et al.* [50]) showed the best prediction for silty sand and fine
 365 sand.

366 According to the requirements of experimental data, comparisons will here be car-
 367 ried out in two groups: models only based on the measured main desorption branch
 368 and models needing additional data besides the main desorption branch.

369 5.1. Models only based on the measured main desorption branch

370 5.1.1. Mualem Model II-1 [51]

371 The Mualem Model II-1 [51] is the extension of Mualem Model II [23]. In a totally
 372 homogeneous porous medium, two pore water distribution functions h and l can be
 373 considered identical for pore necks and pore bodies. This leads to $h(\bar{\rho}) = l(\bar{\rho})$ (or
 374 $h(\bar{r}) = l(\bar{r})$). Hence, the same distribution function is applied to both pore necks and
 375 pore bodies. The water distribution function (see Eq. 1) can be rewritten as:

$$f(\bar{r}, \bar{\rho}) = h(\bar{r})h(\bar{\rho}) \quad (18)$$

376 The water content during the wetting process (see Eq. 4) is rewritten as:

$$\theta_w(P_c) = \theta^{\max} H(P_c) \quad (19)$$

377 This simplification does not affect the expression for water content during a drying
 378 process (see Eq. 5). Combining Eq. (19) with Eq. (6), an equation calculating $\theta_w(P_c)$
 379 from $\theta_d(P_c)$ is derived as:

$$\theta_w(P_c) = \theta^{\max} - [(\theta^{\max})^2 - \theta^{\max}\theta_d(P_c)]^{1/2} \quad (20)$$

380 The equivalent equation as a function of S is:

$$S_w(P_c) = 1 - [1 - S_d(P_c)]^{1/2} \quad (21)$$

381 *5.1.2. Parlange's model [49], modified by Braddock et al. [50]*

382 Based on the Parlange's hysteresis model [49], an expression of the adsorption
383 isotherm can be obtained as the following form.

$$\frac{dS_w}{dP_c} = \frac{S_w - S_d}{P_c} \quad (22)$$

384 Taking VG2 equation to describe the desorption isotherm and integrating Eq. (22),
385 an analytical form of the adsorption isotherm was provided by Braddock *et al.* [50].

$$S_w = -\frac{P_c}{a_{dV}} + \left[1 + \left(\frac{P_c}{a_{dV}} \right)^{n_{dV}} \right]^{1/n_{dV}} \quad (23)$$

386 where a_{dV} and n_{dV} are fitting parameters for the desorption isotherm.

387 *5.2. Models needing additional data besides the main desorption branch*

388 *5.2.1. Feng and Fredlund's model [43], improved by Pham et al. [46]*

389 From section 4, one may know that there are two coefficients in FF model [43] (see
390 Eq. (12)). This indicates that two points on the adsorption isotherm are enough to de-
391 termine these two parameters and to calculate the entire adsorption isotherm. Based on
392 this idea, Pham *et al.* [46] introduced a method to find the positions of two such points
393 (denoted by A and B). The position of point A corresponds to a capillary pressure:

$$P_{c,wA} = \left(\frac{a_d}{10} \right)^{1/b_d} \quad (24)$$

394 where a_d and b_d are parameters used in FF equation [43] for the desorption isotherm.

395 [Figure 7 about here.]

396 Point B on the adsorption isotherm is defined as the point having a capillary pres-
397 sure that is symmetrical to the capillary pressure of point A with respect to a horizontal
398 line passing through point E on the adsorption isotherm (see Fig. 7).

$$|P_{c,wA} - P_{c,wE}| = |P_{c,wB} - P_{c,wE}| \quad (25)$$

399 In the above equation, the position of point E is unknown. Pham *et al.* [46] intro-
 400 duced a method to determine the position of point E. The authors defined two other
 401 points on the desorption isotherm to find point B. One point (C) is in the middle of the
 402 desorption isotherm (at $S = 0.5$):

$$P_{c,dC} = (a_d)^{1/b_d} \quad (26)$$

403 The other point (D) is on the desorption isotherm and has the same saturation value
 404 as point A:

$$P_{c,dD} = \left(\frac{a_d}{S_{wA}} - 1 \right)^{1/b_d} \quad (27)$$

405 Points A, C, D and E should satisfy the relation:

$$|P_{c,wA} - P_{c,wE}| = |P_{c,dD} - P_{c,dC}| \quad (28)$$

406 The capillary pressure of point B on the adsorption isotherm is obtained by solving
 407 Eqs. (25)-(28).

$$P_{c,wB} = P_{c,wA} - 2 \left[\left(\frac{a_d}{S_{wA}} - a_d \right)^{1/b_d} - a_d^{1/b_d} \right] \quad (29)$$

408 Finally, the two parameters (a_w and b_w) used in Feng and Fredlund's model [43]
 409 are calculated on the basis of the capillary pressures at points A and B.

$$a_w = \frac{S_{wA} P_{c,wA}^{b_w}}{1 - S_{wA}} \quad (30)$$

$$b_w = \frac{\log \left[\frac{1 - S_{wB}}{S_{wB}} \frac{S_{wA}}{1 - S_{wA}} \right]}{\log \left[\frac{P_{c,wB}}{P_{c,wA}} \right]} \quad (31)$$

410 5.2.2. *Mualem Model IV [48]*

411 The Mualem Model IV [48] is an improved version of Mualem Model II [23, 51].
 412 In addition to the main desorption branch, a wetting scanning curve is needed to predict
 413 the main adsorption branch. The prediction is divided into two parts: from the highest
 414 capillary pressure to the starting point of the wetting scanning curve, and for capillary
 415 pressure lower than the starting point of the wetting scanning curve. The first part
 416 is predicted using the same method as Mualem Model II-1 [51]. The second part is
 417 calculated by using both the main desorption branch and the wetting scanning curve.
 418 Thus, for the main adsorption branch, the saturation can be expressed as follows:

$$S_w(P_c) = \begin{cases} 1 - [1 - S_d(P_c)]^{1/2} & \text{for } P_{c,1} \leq P_c \leq P_c^{\max} \\ 1 - \frac{1 - S_{1,w}(P_c)}{[1 - S_d(P_{c,1})]^{1/2}} & \text{for } P_c^{\min} \leq P_c < P_{c,1} \end{cases} \quad (32)$$

419 where $P_{c,1}$ is the capillary pressure at the starting point of the additional wetting scan-
 420 ning curve. The subscript (l,w) represents the first wetting scanning curve.

421 The accuracy of this model is sensitive to the position of the starting point of the
 422 wetting scanning curve. If the scanning curve starts at a high RH, the model tends to
 423 be the same as Mualem Model II-1 [51]. If the scanning curve starts at a low RH, it
 424 will only use the wetting scanning curve to predict the adsorption isotherm. According
 425 to the position of the starting point in available experimental data, comparisons were
 426 performed in three levels: low RH ($RH_1 = 12\%$), medium RH ($RH_1 = 33\%$) and
 427 high RH ($RH_1 = 53\%$). Not all materials in the database can reach this requirement.
 428 Only five materials (Concrete1, Concrete2, Concrete3, Paste2 and Paste3) are used for
 429 validation and comparisons.

430 5.3. *Comparison of results and discussion*

431 The prediction results of Paste2 are shown in Figs. 8 (other comparison results are
 432 included in the document “supplementary materials”). For the models only using the
 433 main desorption curve (Mualem Model II-1 [51] and Parlange’s model [49] modified
 434 by Braddock *et al.* [50]), the standard deviation of the statistical criteria R_{adj}^2 and NME
 435 are very large (see Fig. 9). Furthermore, R_{adj}^2 values are low and NME values are

436 high. Figure 8 also displays that the prediction of these two models do not show good
437 agreements with measured curves. As a result, one can conclude that only using one
438 main curve is not sufficient to predict the other main curve in the case of cementitious
439 materials.

440 [Figure 8 about here.]

441 With regard to models needing additional data, Fig. 9 clearly shows that the im-
442 proved Feng and Fredlund's model [43] yields the best prediction with the highest R_{adj}^2
443 and the lowest NME. This is due to the fact that adding two additional measured points
444 significantly improves the prediction accuracy. In spite of needing one wetting scan-
445 ning curve, by contrast, Mualem Model IV [48] does not provide good performance,
446 regardless of the level of RH.

447 [Figure 9 about here.]

448 The requirement of additional experimental data limits the application of the im-
449 proved Feng and Fredlund's model [43]. In fact, the positions of points A and B depend
450 on the shape of the desorption isotherm; thus, they vary according to the studied ma-
451 terials. If these two points are fixed independently of the material, it will be more
452 interesting from a practical point of view to determine the adsorption isotherm. The
453 mean RH values for these two points are calculated according to the prediction results,
454 which are around 67% and 86%. They almost correspond to the RH values (63.2%
455 and 90.4%) used in the measurements of the sorption isotherms in [5]. Thus, these
456 two points can be fixed at 63.2% and 90.4%. The main adsorption isotherms can be
457 determined by using Eq. (12). The predicted results are compared with other methods
458 in Fig. 9. It is clear that the fixed points method has the equivalent accuracy as Feng
459 and Fredlund's model [43] (improved by Pham *et al.* [46]). This implies that if one
460 wants to know the entire adsorption isotherm, it is enough to determine two adsorption
461 points at RH = 63.2% and 90.4%.

462 The above comparisons reveal that it may be not appropriate to predict one main
463 isotherm by using the other main isotherm only (*e.g.*, Mualem Model II-1 [51] and
464 Parlange's model [49] modified by Braddock *et al.* [50]). Two additional measured

465 points on the adsorption isotherm can provide a better prediction (improved Feng and
 466 Fredlund model [43]) than using one wetting scanning curve (Mualem Model IV [48]).

467 **6. Prediction of scanning isotherms**

468 Comparisons of different hysteresis models can be found in the literature [11, 30,
 469 52–54] for a large variety of porous media, not including cementitious materials. By
 470 statistical analysis, Viaene *et al.* [52] concluded that models developed based on do-
 471 main theories give the best predictions. A similar conclusion has been drawn in the
 472 comparisons performed by Pham *et al.* [11]. However, besides the domain models,
 473 several recently developed empirical models were also selected for comparisons due to
 474 their simplicity of application.

475 *6.1. Mualem Model II [23]*

476 In Mualem Model II [23], a series of equations to calculate scanning curves have
 477 been proposed based on Mualem's diagram (see Fig. 2). They use two basic functions
 478 H (see Eq. 3) and L (see Eq. 6), which are determined by two main sorption isotherms
 479 (desorption and adsorption). Rewriting them as a function of S , we get:

$$L(P_c) = S_w(P_c) \quad H(P_c) = \frac{S_d(P_c) - S_w(P_c)}{1 - S_w(P_c)} \quad (33)$$

480 The first scanning curve in drying (scanning curve of order $N = 1$), starting on the
 481 adsorption isotherm at the point $(S_w(P_{c,1}), P_{c,1})$, is formulated according to Mualem's
 482 diagram:

$$S_{1,d}(P_c) = S_w(P_{c,1}) - [L(P_{c,1}) - L(P_c)][1 - H(P_c)] \quad (34)$$

483 The first scanning curve in wetting (order $N = 1$), starting on the desorption
 484 isotherm at the point $(S_d(P_{c,1}), P_{c,1})$, is calculated by:

$$S_{1,w}(P_c) = S_d(P_{c,1}) + [L(P_c) - L(P_{c,1})]H(P_{c,1}) \quad (35)$$

485 It can be noted that the expression for the first drying scanning is different to the
 486 first wetting scanning curve. The expression for the wetting scanning curve of order N
 487 (odd number and > 1) is deduced according to Mualem's diagram:

$$S_{N,w}(P_c) = S(P_{c,N}) + [L(P_c) - L(P_{c,N})][1 - H(P_{c,N})] \text{ for } L(P_c) \leq L(P_{c,N-1}) \quad (36)$$

488 where $S(P_{c,N})$ is the saturation at the starting point of the current scanning curve, which
 489 must be the end point of the previous scanning curve. If $L(P_c) \geq L(P_{c,N-1})$, Mualem's
 490 diagram indicates that $L(P_{c,N-3})$ will be used instead of $L(P_{c,N-1})$ until the scanning
 491 curve reduces to the main isotherms (order $N = 0$). For example in Fig. 10, if the third
 492 scanning curve in wetting CB continues to decrease after point B, it must do so on
 493 curve BD, which is the extension line of the first scanning curve in wetting AB. This
 494 ensures that all scanning loops are closed and enclosed in the main loop.

495 In the same manner, the expression for the drying scanning curve of order N (even
 496 number) is written as:

$$S_{N,d}(P_c) = S(P_{c,N}) - [L(P_{c,N}) - L(P_c)][1 - H(P_c)] \text{ for } L(P_c) \geq L(P_{c,N-1}) \quad (37)$$

497 If $L(P_c) \leq L(P_{c,N-1})$, $L(P_{c,N-3})$ will be used instead of $L(P_{c,N-1})$ for calculation.
 498 Equations (35-37) comprise a set of formulas to calculate arbitrary scanning loops.

499 [Figure 10 about here.]

500 6.2. Mualem dependent model [25–27]

501 Based on the assumption of *dependent* of the neighbouring pores in Mualem Model
 502 III [25], Mualem and Miller [26] and Mualem [27] have provided different weighting
 503 functions p_d to improve the prediction of the scanning curves. The improved version
 504 in [27] seems to provide a better agreement with the experimental data for soils [27, 28,
 505 54]. This version (hereafter called *Mualem dependent model*), as the same as Mualem

506 Model II [23], only needs both main sorption isotherms. The formula of p_d is given
 507 as [27]:

$$p_d(S) = \frac{1 - S}{[1 - S_w(P_c^+)]^2} \quad (38)$$

508 where P_c^+ is the capillary pressure at $S_d(P_c^+) = S$ (the current saturation, see Fig. 11).

509 [Figure 11 about here.]

510 The function for the first scanning curve in drying (order $N = 1$), starting on the
 511 adsorption isotherm, is deduced as:

$$S_{1,d}(P_c) = S_w(P_{c,1}) - p_d(S) [1 - S_w(P_c)] [S_w(P_{c,1}) - S_w(P_c)] \quad (39)$$

512 The first scanning curve in wetting (order $N = 1$), starting on the main desorption
 513 isotherm, can be calculated by:

$$S_{1,w}(P_c) = S_d(P_{c,1}) + p_d(S_1) [1 - S_w(P_{c,1})] [S_w(P_c) - S_w(P_{c,1})] \quad (40)$$

514 where $p_d(S_1)$ is p_d at the starting point $(S_d(P_{c,1}), P_{c,1})$. Functions for scanning loops
 515 can be easily derived based on Mualem's diagram [25–27]. For the wetting scanning
 516 curves of order N (odd number and > 1), the expression is written as:

$$S_{N,w}(P_c) = S(P_{c,N}) + p_d(S_N) [1 - S_w(P_{c,N})] [S_w(P_c) - S_w(P_{c,N})], S_w(P_c) \leq S_w(P_{c,N-1}) \quad (41)$$

For the drying scanning curves of order N (even number), the expression is:

$$S_{N,d}(P_c) = S(P_{c,N}) - p_d(S) [1 - S_w(P_c)] [S_w(P_{c,N}) - S_w(P_c)], S_w(P_c) \geq S_w(P_{c,N-1}) \quad (42)$$

517 One can see that wetting scanning curves are provided by explicit formulas, while
 518 equations for drying scanning curves are implicit forms with respect to the current
 519 capillary pressure P_c . Therefore, a drying scanning curve must be calculated by an
 520 iterative method. Substitution of Eq. (38) into Eq. (40) yields the same equation for

521 the first scanning curve in wetting as the one used by Mualem Model II. It means that
 522 these two models will share the same first wetting scanning curve.

523 6.3. Incremental models

524 Elastoplastic models have been used in porous media field to describe the rela-
 525 tion between P_c and S [32, 33]. Usually, a simple incremental relation is used. It is
 526 generally assumed that the differential of P_c on the scanning curve is related to the
 527 differential of P_c or S on the main isotherm. Two typical models are described below.

528 (1) Li's model [32]

529 In this method, the scanning curve is obtained by integrating the incremental re-
 530 lation $dP_c - dS$. A proposition for the incremental form of $P_c(S)$ is given in [32] as
 531 follows:

$$d \ln[P_c(S)] = \left[\frac{\ln[\bar{P}_c(S)] - \ln(P_{c,1})}{\ln[P_c(S)] - \ln(P_{c,1})} \right]^\beta d \ln[\bar{P}_c(S)] \quad (43)$$

532 where β is a material parameter. $\bar{P}_c(S) = P_{c,d}(S)$ is the capillary pressure on the main
 533 desorption branch for the drying process and $\bar{P}_c(S) = P_{c,w}(S)$ is on the main adsorption
 534 branch for the wetting process. By integrating Eq. (43), one can obtain the expressions
 535 for a drying scanning curve,

$$\ln[P_c(S)] = \ln(P_{c,1}) + \left(|\ln[P_{c,d}(S)] - \ln(P_{c,1})|^{\beta_d+1} - |\ln(P_{c,d1}) - \ln(P_{c,1})|^{\beta_d+1} \right)^{1/(\beta_d+1)} \quad (44)$$

536 and for a wetting scanning curve.

$$\ln[P_c(S)] = \ln(P_{c,1}) - \left(|\ln(P_{c,1}) - \ln[P_{c,w}(S)]|^{\beta_w+1} - |\ln(P_{c,1}) - \ln(P_{c,w1})|^{\beta_w+1} \right)^{1/(\beta_w+1)} \quad (45)$$

537 One can notice that in this method the scanning curve always starts with an infinite
 538 slope (*i.e.*, $|dP_c/dS| \rightarrow \infty$).

539 (2) Wei's model [33]

540 Based on reference [55], Wei *et al.* [33] suggested a new and simpler incremental
 541 relation, expressed as:

$$dP_c = -K_p dS \quad (46)$$

542 where $-K_p$ is the slope of the scanning curve.

543 According to Wei's theory, when P_c changes, the water content in the material is as-
 544 sumed to be additively decomposed into two parts, a reversible part and an irreversible
 545 part. The irreversible part of water content is dependent on the capillary pressure, while
 546 the reversible part is independent. Accordingly, the slope of each scanning curve can
 547 be divided into two parts, one is an irreversible part affected by the main isotherm
 548 (*e.g.*, the first term on the right side of Eq. 47), and the other is based on the effect of
 549 both reversible and irreversible water content (*e.g.*, the second term on the right side of
 550 Eq. 47). At current P_c , the slope of the drying scanning curve could be calculated by:

$$K_p = -\frac{dP_{c,d}(S)}{dS} + \frac{c_d |P_c - P_{c,d}(S)|}{P_{c,d}(S) - P_{c,w}(S) - |P_c - P_{c,d}(S)|} \quad (47)$$

For a wetting scanning curve, it is:

$$K_p = -\frac{dP_{c,w}(S)}{dS} + \frac{c_w |P_c - P_{c,w}(S)|}{P_{c,d}(S) - P_{c,w}(S) - |P_c - P_{c,w}(S)|} \quad (48)$$

551 where c_d and c_w are the internal state variables for drying and wetting processes, re-
 552 spectively. In Eqs. (47) and (48), the slope of the main sorption isotherm ($-\frac{dP_{c,d}(S)}{dS}$)
 553 and the distance between two main isotherms ($P_{c,d}(S) - P_{c,w}(S)$) represent irreversible
 554 effects. The part of $|P_c - P_{c,d}(S)|$, depending on the current capillary pressure, repre-
 555 sents the reversible effect. The coefficients of c_d (or c_w) is used to control the reversible
 556 effect and it is a function of the water content. To simplify the application of this model,
 557 Wei *et al.* [33] assumed constant values of c_d and c_w . Clearly, the reversible effect can
 558 be observed at the beginning of the scanning curve due to the large initial slope, like it
 559 has been illustrated by Wei *et al.* [33] and in the model of Hogarth *et al.* [56, 57] for
 560 soils.

561 When Eqs. (47) and (48) are applied to simulate scanning curves for cementitious

562 materials, the model predict that the scanning curve crosses the main sorption isotherm
 563 because of the significant reversible effect. However, experimental data of the scan-
 564 ning curves [5, 36] do not show such a reversible effect. Hence, a modification to
 565 Eqs. (47) and (48) is proposed in the present paper to reduce the reversible effect and
 566 new relations are yielded for a drying scanning curve:

$$K_p = -\frac{dP_{c,d}(S)}{dS} + \frac{c_d|P_c - P_{c,d}(S)|}{P_{c,d}(S) - P_{c,w}(S)} \quad (49)$$

and for a wetting scanning curve.

$$K_p = -\frac{dP_{c,w}(S)}{dS} + \frac{c_w|P_c - P_{c,w}(S)|}{P_{c,d}(S) - P_{c,w}(S)} \quad (50)$$

567 One can notice that K_p depends on the current position of the scanning curve. An
 568 iterative method is needed to compute the current capillary pressure.

569 6.4. Improved Rubin's empirical hysteresis model [58]

570 The basic idea for the calculation of the scanning curve is to determine the dis-
 571 tance between the scanning curve and the main isotherm which the scanning curve
 572 approaches. Here, considering the saturation as an independent variable, the present
 573 paper takes the prediction of the drying scanning curve as an example of how to de-
 574 velop an empirical hysteresis model.

575 As above models show, the distance between the drying scanning curve and the
 576 main desorption curve should be a function of the current saturation (S_d) and the start-
 577 ing point of the scanning curve, which is simply formulated as:

$$P_{c,d} - P_c = F(S_d, P_{c,1}, S_1) \quad (51)$$

578 Various F functions can be found in the literature. For instance, Rubin [58], as
 579 well as Feng and Fredlund [43], proposed expressions for the first drying scanning
 580 curve, considering F as proportional to the distance between the two main isotherms.
 581 Nevertheless, these two models did not take into account the position of the starting
 582 point in their expressions. Thus, these expressions were only valid for the first scanning

583 curve in drying and not able to simulate scanning loops. Here, a new F is introduced by
 584 adapting the original form in [58]. The exponential relation is used to control how fast
 585 the scanning curve approaching the main isotherm. Finally, the following expression
 586 is given for the drying scanning curve:

$$P_c(S) = P_{c,d}(S) - [P_{c,d}(S) - P_c(S_1)] \exp[\gamma_d(S - S_1)] \quad (52)$$

587 The negative value of $\gamma_d(S - S_1)$ limits the exponential expression from 1 (at the
 588 starting point) to near 0 (almost on the main isotherm). Similarly, the expression for
 589 the wetting scanning curve is derived as:

$$P_c(S) = P_{c,w}(S) + [P_c(S_1) - P_{c,w}(S)] \exp[\gamma_w(S_1 - S)] \quad (53)$$

590 The constant γ_d and γ_w are used to determine the shape of the scanning curve and
 591 needs to be assessed from experimental data.

592 6.5. Comparison of results and analysis of hysteresis models

593 6.5.1. Predictions of the first scanning curves

594 Overall, five materials, Concrete1, Concrete2, Concrete3, Paste2 and Paste3 (see
 595 Table 2), have been studied. One example of a comparison between experimental data
 596 and predictions (Paste3) performed by above models are provided in Fig. 12 (other
 597 comparison results are included in the document “supplementary materials”). The re-
 598 sults obtained by Mualem Model II [23] and Mualem dependent model [27] are illus-
 599 trated in the same figure. For the empirical models with unknown parameters (β_d and
 600 β_w in Eqs. 44 and 45, c_d and c_w in Eqs. 49 and 50 and γ_d and γ_w in Eqs. 52 and 53), the
 601 models have been optimized to predict all measured wetting scanning curves for each
 602 material.

603 [Figure 12 about here.]

604 The empirical models reveal that the predicted wetting scanning curves increase
 605 slowly at the beginning where the curves are almost parallel to the horizontal axis (see
 606 Fig. 12), so that these curves can approach the main adsorption branch quickly. This

607 has been observed by Åhs [36] in experimental results on the hardened cementitious
608 materials. As a consequence, the first wetting scanning curves probably reach the
609 main curve before $RH = 100\%$, as illustrated by modified Wei's model and improved
610 Rubin's model (see Fig. 12). The wetting scanning curves predicted by Mualem's
611 models [23, 27] increase more smoothly and reach the main wetting curve at RH close
612 to 100% . This results in a poorer agreement with measured data than empirical models.

613 Experimental data for the first drying scanning curves are only available for Paste2
614 and Paste3 materials in our database [5]. Predictions from Mualem Model II [23] show
615 that curves decrease with a lower slope and turn towards the desorption isotherm more
616 gently than curves predicted by the empirical models. Therefore, the empirical models
617 yield predictions with a lower accuracy compared with Mualem Model II. With respect
618 to Mualem dependent model [27], it does not provide a good prediction for the drying
619 scanning curves (see Fig. 12a).

620 The two criteria R_{adj}^2 and NME for the prediction of the first wetting scanning
621 curves are shown in Fig. 13. They illustrate that the empirical models can give higher
622 R_{adj}^2 and lower NME values than the Mualem Model II. This is definitely due to that
623 the two additional parameters in the empirical models can be adjusted to fit the exper-
624 imental data, while Mualem's models essentially depend on the shape of the two main
625 sorption isotherms. In contrast, in the prediction of the first drying scanning curves for
626 Paste3, Mualem Model II reveals better prediction than the empirical models.

627 Figure 13 also shows that all empirical models have almost equal accuracy for the
628 estimation of the first scanning curves, either wetting or drying. None of the selected
629 hysteresis models is considerably better than the others. This conclusion is similar to
630 that reported by Jaynes [30] when he compared hysteresis models for soils. Hence, it
631 needs the comparison of higher order scanning curves.

632 [Figure 13 about here.]

633 6.5.2. Predictions of scanning loops

634 Scanning loops have been measured for Paste2 and Paste3 [5]. Here three measured
635 loops, one for Paste2 and two for Paste3, have been chosen to verify the applicabilities

636 of the selected hysteresis models. One of the prediction result is shown in Fig. 14 (other
637 comparison results are included in the document “supplementary materials”).

638 Because of the poor prediction of the first wetting scanning curves, Mualem de-
639 pendent model [27] do not show better prediction results than Mualem Model II [23].
640 Among the empirical models, Li’s model [32] appears to be the best one to predict
641 the shape of the measured scanning loops. Besides, the improved Rubin’s model [58]
642 shows a better prediction of secondary drying scanning curves than the modified Wei’s
643 model [33] (see Fig. 14).

644 [Figure 14 about here.]

645 6.5.3. Analysis of the pumping effect

646 One hysteretic behavior, called the *pumping effect*, should be noticed for the em-
647 pirical models. It refers to the non-closure of scanning loops when the secondary and
648 higher order scanning curves are calculated. Researchers consider that the pumping
649 effect is an artifact of the algorithm and does not correspond to an actual material prop-
650 erty [57, 59]. Thus, this behavior is also called the *pumping errors* (PEs), which is
651 defined and illustrated in Fig. 15. Point A is the starting point of the scanning loop.
652 Point B is on the secondary scanning curve in drying at the same RH as point A.
653 The difference of S between A and B is defined as the *drying pumping error* (DPE).
654 Similarly, a *wetting pumping error* (WPE), originating from wetting scanning curves
655 (difference between point C and point D in Fig. 15), is also observed for the empirical
656 models. Figure 15 implies the amplitudes of DPE and WPE can be quantified.

657 [Figure 15 about here.]

658 The evolution of PE is not only dependent upon the initial state but also upon the
659 magnitude of the RH oscillations. Thus, to assess the evolution of DPE and WPE, two
660 initial states of RH cycles are tested: an initial drying state and an initial wetting state.
661 Paste3 material is taken as an example because many measured scanning curves are
662 available for this material, including the first and secondary drying curves and the first
663 wetting scanning curves. Two RH oscillations have been chosen for each initial state.
664 The first one has been provided by experiments, with RH changing between 63% and

665 97%. The second one reduces the RH difference to 20%, such as for an RH oscillating
666 between 63% and 83%.

667 The simulated results of PEs *vs.* the number of cycles are plotted in Fig. 16. The
668 PEs are clearly more significant for the small RH oscillations than for the large RH
669 oscillations. The first loop produces the largest PE, and then it reaches a constant
670 value after about 2 cycles for the large RH oscillations and about 5 cycles for the small
671 RH oscillations. Li's [32] and improved Rubin's models [58] provide similar PE (less
672 than 0.1) for the large RH oscillations, while improved Rubin's model yields larger PE
673 (around 0.3) than Li's model (around 0.2) for small RH oscillations. It is also clear
674 that WPEs are larger than DPEs in both empirical models for the same kind of the RH
675 oscillations.

676 [Figure 16 about here.]

677 For the large RH changes, PEs of the empirical models are acceptable. Neverthe-
678 less, for the small RH oscillations, if the model fails to eliminate PEs, the cumulative
679 errors associated with oscillations of hysteresis loops are probably significant and lead
680 to unrealistically simulated results. Some works have been done to avoid PEs when
681 a hysteresis model was developed. By assuming that scanning curves have the same
682 shape as the main branches, Parker and Lenhard [60] and Huang *et al.* [57] enforced
683 closure of scanning loops to eliminate pumping errors. The problem of these methods is
684 that they need to rescale and reformulate each scanning curve if the direction changes,
685 and this may be inefficient for modelling of frequent drying and wetting cycles.

686 6.6. Discussion

687 Compared with the empirical models, Mualem's models (Model II [23] and depen-
688 dent model [27]) have the advantage to avoid PE, even though their predictions are not
689 as good as the empirical models. They may be more applicable for smaller RH oscilla-
690 tions comparing with the empirical models. Another advantage of Mualem's models is
691 that they need only the two main sorption isotherms to predict scanning loops, unlike
692 the empirical models that require at least two scanning curves to determine parameters.

693 Notably, from a practical point of view, equations of Mualem Model II [23], Li's
694 model [32] and improved Rubin's model are expressed in explicit forms which are
695 easier to implement in a numerical modelling of moisture transport. In particular, the
696 improved Rubin's model uses very simple equations and provides an acceptable pre-
697 diction. The modified Wei's model [33] and Mualem dependent model [27] correspond
698 to implicit formulas which significantly increase the difficulty to apply them.

699 To conclude the above analysis about hysteresis models, this paper recommends
700 Mualem Model II [23], Li's [32] and improved Rubin's models in the case of moisture
701 transport modelling within cementitious materials exposed to cyclic boundary condi-
702 tions. Two empirical models (Li's [32] and improved Rubin's models) are more suit-
703 able for large RH oscillations, in which pumping effects can be reduced.

704 **7. Conclusions and recommendations**

705 Detailed comparisons of hysteresis models for cementitious materials have been
706 done in this study. Experimental results measured by the saturated salt solution method
707 have been used for the validation and verification of models. Two statistical criteria
708 R_{adj}^2 and NME have been employed to evaluate models and their relevance. In regard
709 to the proposed multi-level hysteresis modelling presented in Table 1, above studies
710 indicate that Level 1 and Level 2 provide more reliable prediction results. Because
711 there is no model that can predict one main curve well only on the basis of the other
712 main curve, the prediction accuracy of Level 3 is not satisfactory. Evaluations for
713 all selected models, the main features, disadvantages, and our recommendations, are
714 summarised in the document "supplementary materials". Conclusions can be drawn
715 from above comparisons.

- 716 1. Even though cementitious materials have quite different pore structures com-
717 pared with soils and sands, most selected models in the present paper can pro-
718 vide a satisfactory fitting or prediction in the case of cementitious materials.
- 719 2. Concerning the fitting of the main branches, three-parameter fitting models pro-
720 vide better results than two-parameter models. That means if one wants to pre-

721 dict both main sorption isotherms, at least three points for each isotherm have
722 to be measured.

723 3. The models selected for predicting one main isotherm from the other main
724 isotherm are not satisfactory. However, by using two additional points on the
725 adsorption isotherm, the Fend and Fredlund's model [43] (improved by Pham *et*
726 *al.* [46]), shows a high prediction accuracy. In the present paper, it is found
727 that if these two points are fixed at RH = 63.2% and 90.4%, the model still
728 provides very good prediction. This result could be used to determine the main
729 adsorption branch based on the experimental data at these two points.

730 4. Among the hysteresis models, the empirical ones provide better predictions but
731 pumping errors cannot be avoided with these models. In contrast, Mualem's
732 models show no pumping errors, but predictions do not agree with experimental
733 data as well as the empirical models. It indicates two potential ways to develop a
734 new hysteresis model: one is to improve the accuracy of the conceptual models;
735 the other way is to develop empirical models without pumping errors.

736 This study has investigated the use of hysteresis models to fit the main sorption
737 isotherms and to predict the scanning loops. The purpose is to recommend hystere-
738 sis models for modelling of moisture transport in cementitious materials under cyclic
739 drying and wetting conditions. In the future, the implementation of selected hysteresis
740 models into moisture transport models will provide further comparisons with experi-
741 mental data, such as water content profiles (*e.g.*, gamma-ray attenuation profiles and
742 NMR profiles). Furthermore, these models will be coupled with the available ions
743 transport models [61, 62] and to assess durability of concrete structures exposed to
744 various environments.

745 **Acknowledgement**

746 The research leading to these results has received funding from the European Union
747 Seventh Framework Programme (FP7 / 2007-2013) under grant agreement 264448.

748

749 **Nomenclature**

750	$\bar{\rho}$	Normalized radii of pore bodies	-
751	$\bar{P}_c(S)$	The capillary pressure on the main sorption isotherm used in the Li's model	Pa
752	\bar{r}	Normalized radii of pore necks	-
753	β_d, β_w	Parameters used in the Li's model	-
754	γ_d, γ_w	Parameters used in the improved Rubin's model	-
755	ρ_l	Liquid-water density	$\text{kg} \cdot \text{m}^{-3}$
756	σ	The standard deviation of log-transformed pore radius	-
757	θ	Water content	$\text{m}^3 \cdot \text{m}^{-3}$
758	θ^{\max}	The maximum water content	$\text{m}^3 \cdot \text{m}^{-3}$
759	θ_r	Water content at dry reference state	$\text{m}^3 \cdot \text{m}^{-3}$
760	a_F	Parameters in the FF model	Pa
761	a_F	Parameters in the FX model	Pa
762	a_V	Parameters in the van Genuchten's model	Pa
763	a_{dV}	Parameters for desorption isotherm in the Parlange's model	Pa
764	a_d	Parameters for the main desorption branch in the Feng and Fredlund's model	Pa
765			Pa
766	a_w	Parameters for the main adsorption branch in the Feng and Fredlund's model	Pa
767			Pa
768	b_F	Parameters in the FF model	-
769	b_d	Parameters for the main desorption branch in the Feng and Fredlund's model	-

770	b_w	Parameters for the main adsorption branch in the Feng and Fredlund's model	-
771	c_d, c_w	Parameters used in the Wei's model	-
772	$f(\bar{r}, \bar{\rho})$	The total pore water distribution function	-
773	$h(\bar{r})$	The pore water distribution function related to pore necks	-
774	$H(P_c)$	The cumulative pore water distribution function related to pore necks	-
775	K_p	The slope of a scanning curve	-
776	$l(\bar{\rho})$	The pore water distribution function related to pore bodies	-
777	$L(P_c)$	The cumulative pore water distribution function related to pore bodies	-
778	m	The number of undetermined parameters in a model	-
779	m_F, n_F	Parameters in the FX model	-
780	M_v	The molar mass of water molecule	$\text{kg} \cdot \text{mol}^{-1}$
781	m_V, n_V	Parameters in the van Genuchten's model	-
782	n	The number of measured data	-
783	n_{dV}	Parameters for desorption isotherm in the Parlange's model	-
784	P_c	Capillary pressure	Pa
785	p_d	A weighting factor used for the drying process in Mualem dependent model	-
786	p_w	A weighting factor used for the wetting process in Mualem dependent model	-
787	$P_{c,1}$	Capillary pressure at the first point of a scanning curve	Pa
788	$P_{c,d1}$	Capillary pressure on the main desorption branch at the same saturation as the	
789		first point of a scanning curve	Pa
790	$P_{c,dA}$	Capillary pressure at point A on the main desorption branch	Pa
791	$P_{c,d}$	Capillary pressure of the main desorption branch	Pa

792	$P_{c,w1}$	Capillary pressure on the main adsorption branch at the same saturation as the	
793		first point of a scanning curve	Pa
794	$P_{c,wA}$	Capillary pressure at point A on the main adsorption branch	Pa
795	$P_{c,w}$	Capillary pressure of the main adsorption branch	Pa
796	P_c^+	The capillary pressure used to calculate $p_d(S)$ in the Mualem dependent model	
797			Pa
798	Q	The complementary normal distribution function	-
799	R	The gas constant	$J \cdot K^{-1} \cdot mol^{-1}$
800	R^2	The coefficient of determination	-
801	R_{adj}^2	The adjusted R^2	-
802	RH	Relative humidity	-
803	RH_1	RH at the first point of a scanning curve	-
804	S	Saturation	-
805	$S(P_{c,N})$	Saturation at the starting point of the current scanning curve	-
806	$S_{1,d}(P_c)$	Saturation of the first scanning curve in drying	-
807	$S_{1,w}(P_c)$	Saturation of the first scanning curve in wetting	-
808	S_1	Saturation at the first point of a scanning curve	-
809	S_d	Saturation of the main desorption branch	-
810	$S_{N,d}(P_c)$	Saturation of the N th scanning curve in drying	-
811	$S_{N,w}(P_c)$	Saturation of the N th scanning curve in wetting	-
812	S_w	Saturation of the main adsorption branch	-
813	T	Temperature	K

814	w_i	weighting factor for pore system i in the multi-modal model	-
815	P_c^{\max}	The maximum capillary pressure	Pa
816	P_c^{\min}	The minimum capillary pressure	Pa
817	P_{cm}	Capillary pressure related to the medium pore radius	Pa
818	NME	The normalized mean error	-
819	Subscripts		
820	d	Drying or desorption process	-
821	w	Wetting or adsorption process	-

822 **Appendix A. Supplementary materials**

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Table 1: Proposed multi-level models of hysteresis.

Levels	Available experimental data	Description
Level 1	Both main branches and one scanning curve for each drying and wetting process	One scanning curve is used to determine parameters.
Level 2	Both main branches	Predicting scanning loops without additional parameters
Level 3	One main branch	Determining the other main branch and then predicting scanning loops

Table 2: Collected datasets of WVSIs measured by the saturated salt solution method [5].

No.	Name	Materials	Binders	W/C (W/B)
1	Concrete1	Concrete	OPC	0.45
2	Concrete2	Concrete	OPC+10%SF ^a	0.27 (0.24)
3	Concrete3	Concrete	OPC	0.43
4	Paste1	Paste	OPC	0.45
5	Paste2	Paste	OPC+10%SF ^a	0.20 (0.18)
6	Paste3	Paste	OPC	0.35
7	Paste4	Paste	OPC	0.45
8	Paste5	Paste	OPC	0.60

^aSilica fume.

980 **List of Figures**

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1008		B.	52
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1010		lund’s model [43] (improved by Pham <i>et al.</i> [46]) are at RH = 67% and	
1011		RH = 46%.	53
1012	9	Comparison results of predicting the main adsorption branch from the	
1013		main desorption branch. Results are from models: Mualem Model II-	
1014		1 (MII-1) [51], Parlange’s model (P) [49] (modified by Braddock <i>et</i>	
1015		<i>al.</i> [50]), Feng and Fredlund model’s (FF) [43] (improved by Pham <i>et</i>	
1016		<i>al.</i> [46]), fixed points version of Feng and Fredlund’s model (FF-f)	
1017		and Mualem Model IV [48] of the wetting scanning curve starting at	
1018		RH ₁ =12% (MIV1), RH ₁ =33% (MIV2) and RH ₁ =53% (MIV3). . . .	54
1019	10	An example of scanning loops calculated by Eqs. (35)-(37).	55
1020	11	The calculation of p_d during a drying process in Mualem dependent	
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1022	12	Predicted wetting scanning curves (solid lines) and drying scanning	
1023		curves (dashed-dotted lines) for Paste3 compared with experimental	
1024		data (symbols) from [5].	57

1025	13	Comparison results for wetting scanning isotherm predicted by Mualem	
1026		Model II (M-II) [23], Li's model (L) [32], modified Wei's model (W-	
1027		M) [33] and improved Rubin's model (Im.-Rubin) [58].	58
1028	14	Predicted scanning loop (solid lines) for Paste3 compared with experi-	
1029		mental data (symbols) from [5]. Simulated RH cycle: 97% → 53% →	
1030		99%.	59
1031	15	Definitions of DPE and WPE.	60
1032	16	PE evolutions calculated based on two RH fluctuations.	61

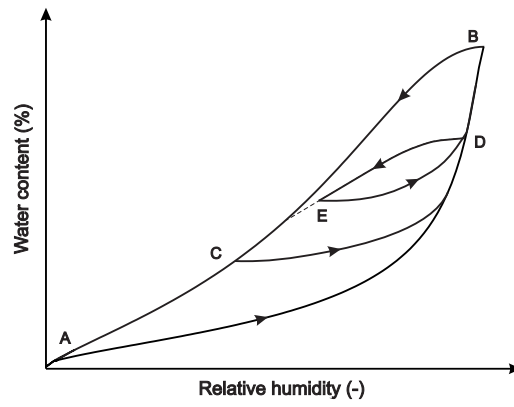
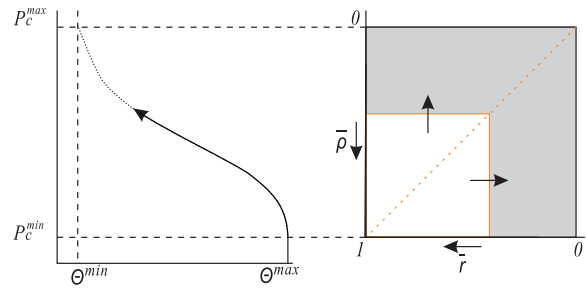
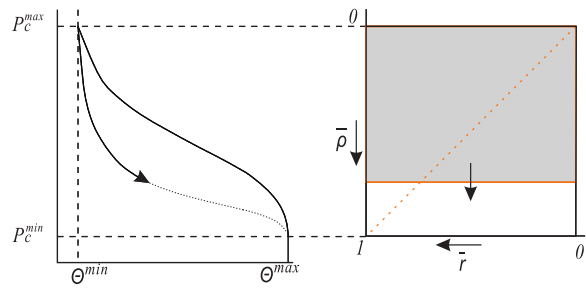


Figure 1: Schematic representation of WVSIs. The main desorption curve (*BA*), the main adsorption curve (*AB*), the first scanning curve in drying (*DE*), the first scanning curve in wetting (*CB*) and the second scanning curve in wetting (*EB*) are illustrated.



(a) The desorption process



(b) The adsorption process

Figure 2: Schematic representation of Mualem's diagram [23]. The gray domain represents completely-filled pores and the blank domain stands for fully-dried pores.

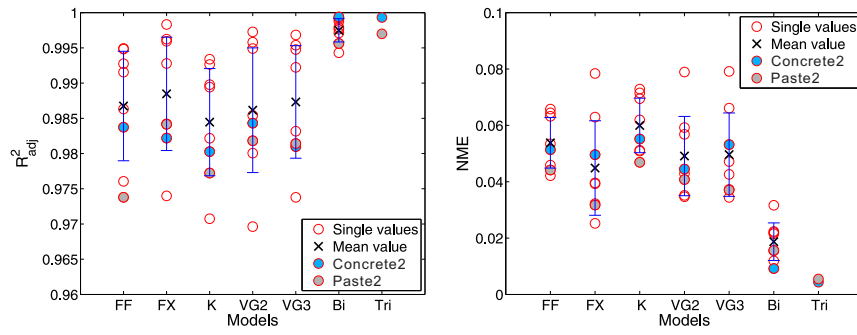


Figure 3: Comparison of R^2_{adj} and NME for desorption isotherms. Each circle (o) in the figure stands for one material (see Table 2). Materials Concrete2 and Paste2 are highlighted. The error bars represent the standard deviation for each studied statistical criteria (R^2_{adj} and NME) and each model.

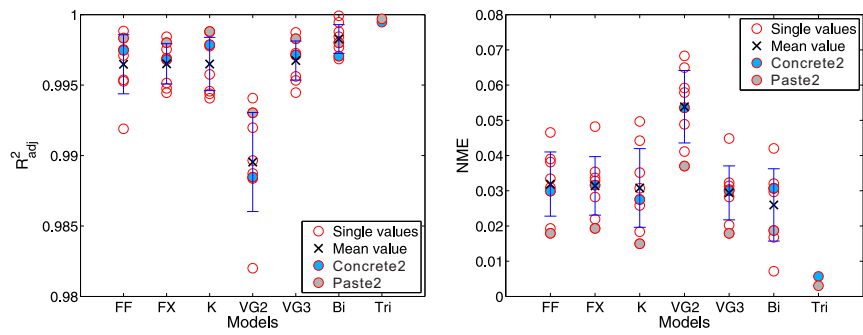
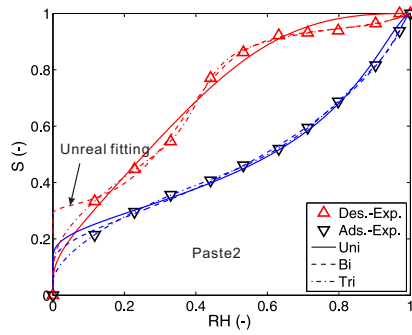
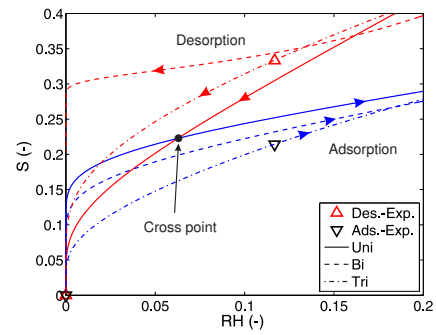


Figure 4: Comparison of R^2_{adj} and NME for adsorption isotherms. Each circle (o) in the figure stands for one material (see Table 2). Materials Concrete2 and Paste2 are highlighted. The error bars represent the standard deviation for each studied statistical criteria (R^2_{adj} and NME) and each model.

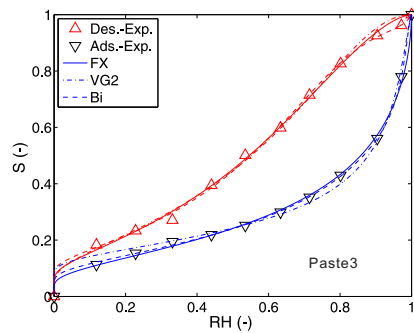


(a) Paste2 material.

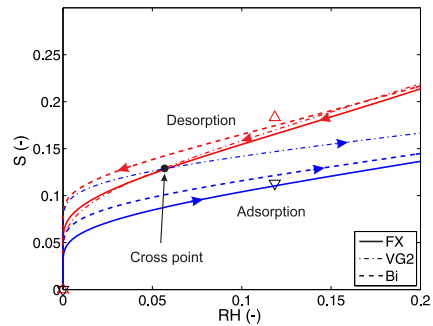


(b) Enlarged figure for Paste2.

Figure 5: Fitting results for Paste2 material to compare results of uni-(VG2), bi- and tri-modal models. Fitting results for low RH values are shown in enlarged figures.



(a) Paste3 material.



(b) Enlarged figure for Paste3.

Figure 6: Fitting results for Paste3 material. To make the display clearly, only three models, VG2, FX and bi-modal models, are chosen to represent uni- and multi-modal models. Fitting results for low RH values are shown in the enlarged figure.

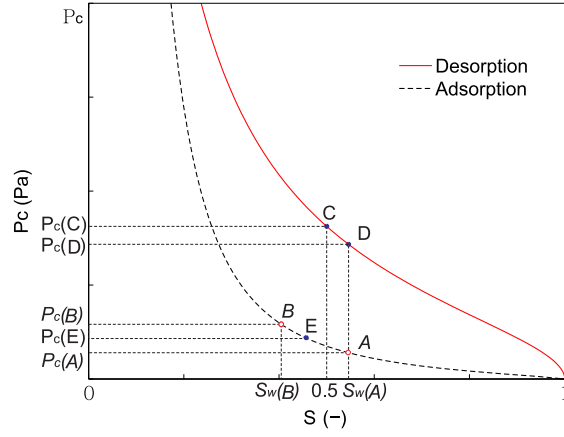
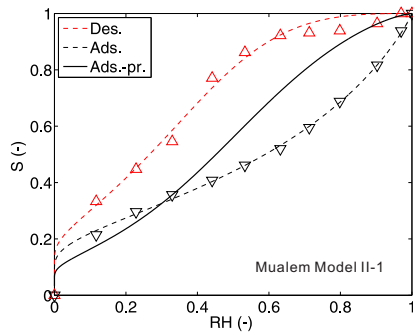
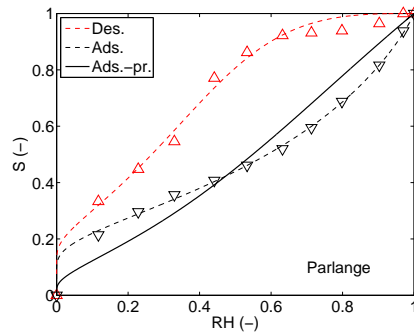


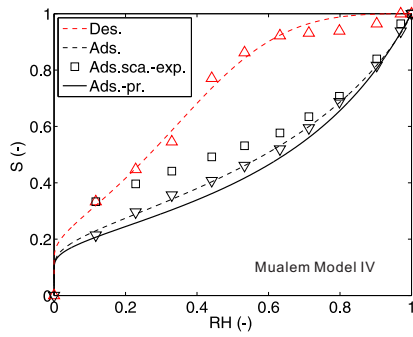
Figure 7: Schematic drawing of Feng and Fredlund's model [43] (improved by Pham *et al.* [46]) used to predict the adsorption branch from the desorption branch. Three points A, D and C are used to calculate point B.



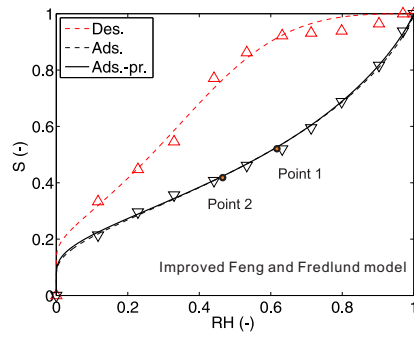
(a) Mualem Model II-1 [51].



(b) Parlange's model [49] (modified by Brad-dock *et al.* [50]).



(c) Mualem Model IV [48], $RH_1=12\%$.



(d) Feng and Fredlund's model [43] (improved by Pham *et al.* [46]).

Figure 8: Prediction results for Paste2 material. Two points in the Feng and Fredlund's model [43] (improved by Pham *et al.* [46]) are at $RH = 67\%$ and $RH = 46\%$.

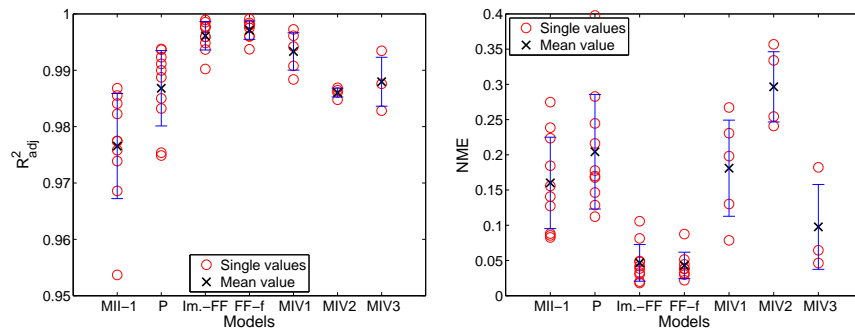


Figure 9: Comparison results of predicting the main adsorption branch from the main desorption branch. Results are from models: Mualem Model II-1 (MII-1) [51], Parlange’s model (P) [49] (modified by Brad-dock *et al.* [50]), Feng and Fredlund model’s (FF) [43] (improved by Pham *et al.* [46]), fixed points version of Feng and Fredlund’s model (FF-f) and Mualem Model IV [48] of the wetting scanning curve starting at $RH_1=12\%$ (MIV1), $RH_1=33\%$ (MIV2) and $RH_1=53\%$ (MIV3).

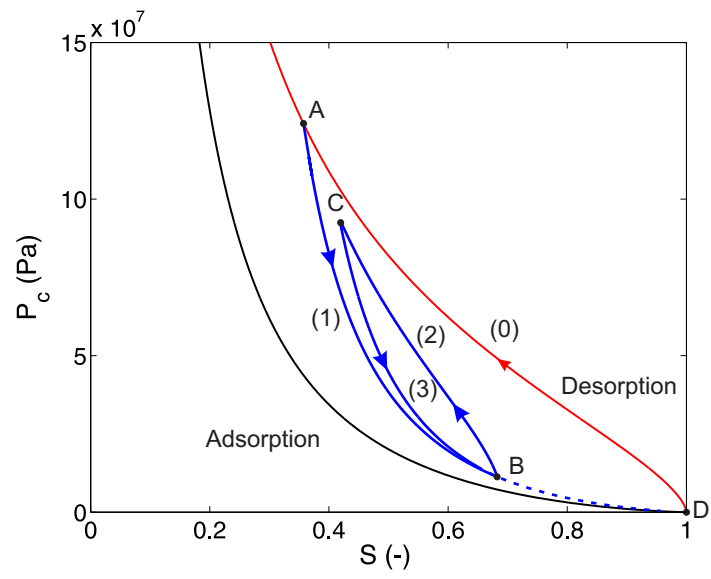


Figure 10: An example of scanning loops calculated by Eqs. (35)-(37).

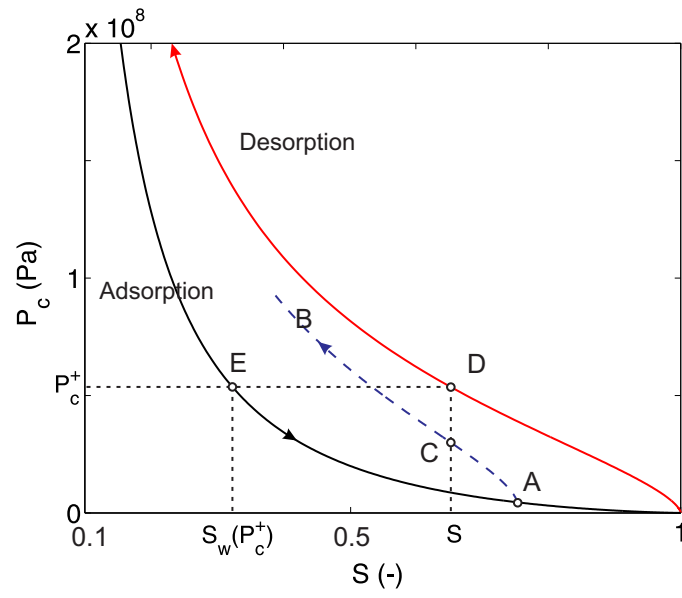
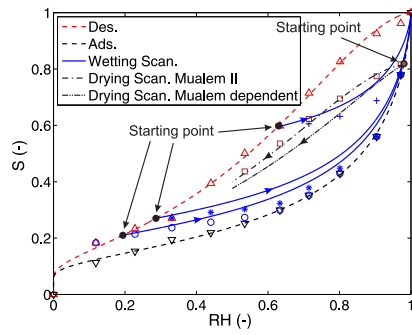
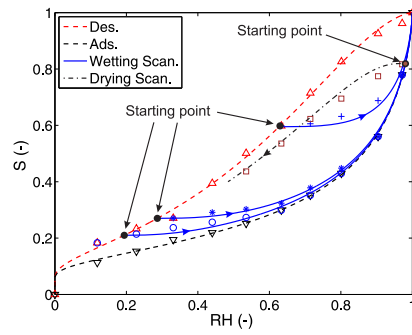


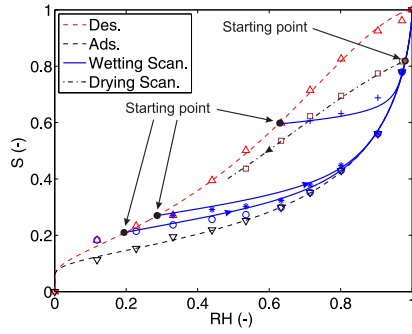
Figure 11: The calculation of p_d during a drying process in Mualem dependent model [27].



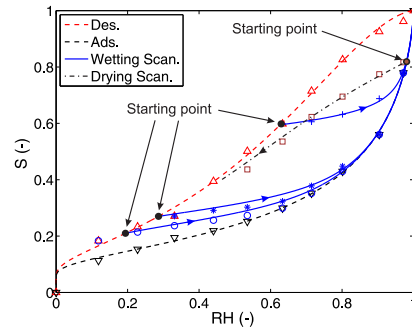
(a) Mualem's models [23, 27].



(b) Li's model [32].



(c) Modified Wei's model (proposed).



(d) Improved Rubin's model (proposed).

Figure 12: Predicted wetting scanning curves (solid lines) and drying scanning curves (dashed-dotted lines) for Paste3 compared with experimental data (symbols) from [5].

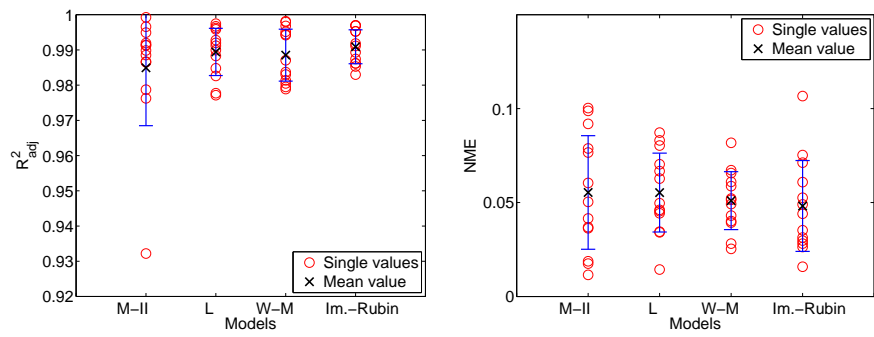
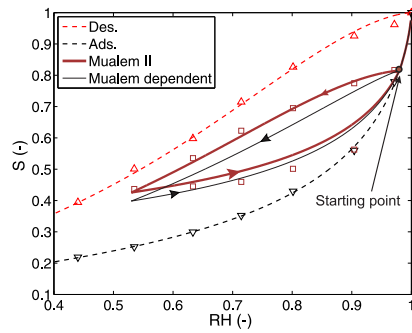
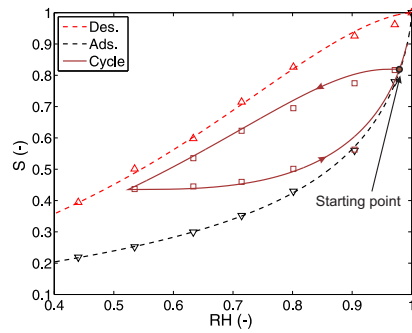


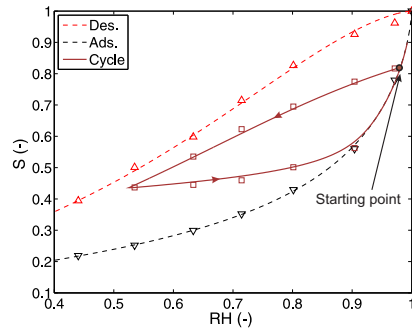
Figure 13: Comparison results for wetting scanning isotherm predicted by Mualem Model II (M-II) [23], Li's model (L) [32], modified Wei's model (W-M) [33] and improved Rubin's model (Im.-Rubin) [58].



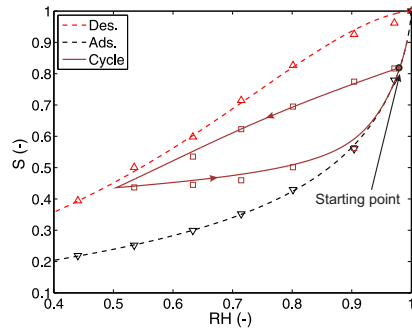
(a) Mualem's models [23, 27].



(b) Li's model [32].



(c) Modified Wei's model (proposed).



(d) Improved Rubin's model (proposed).

Figure 14: Predicted scanning loop (solid lines) for Paste3 compared with experimental data (symbols) from [5]. Simulated RH cycle: 97% \rightarrow 53% \rightarrow 99%.

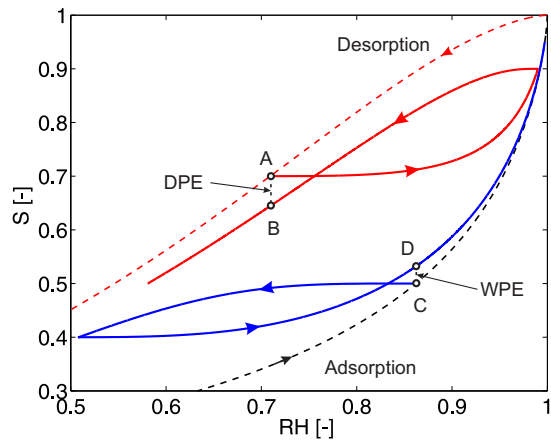
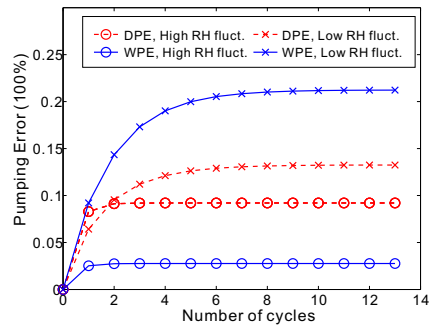
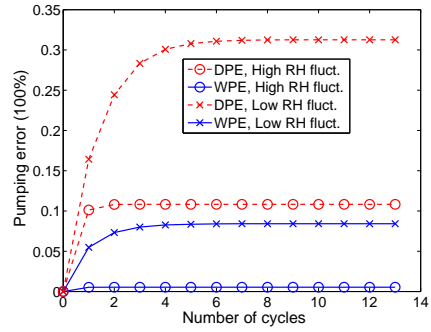


Figure 15: Definitions of DPE and WPE.



(a) The Li's model [32].



(b) The improved Rubin's model.

Figure 16: PE evolutions calculated based on two RH fluctuations.